## Fingerprinting and Forensic Techniques for Landfill Gas Geochemical Assessment

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## Orange County LEA Sam Abu-Shaban

Senior Civil Engineer 1241 E. Dyer Road, Suite 120 Santa Ana, CA 92705 (714) 433-6271



## CIWMB Abel M. Centeno

Chemical Engineer 1001 I Street MS #20 Sacramento, CA 95814 (916) 341-6724

## Session Agenda

#### Theory of Gas Geochemistry

- Introduction Stating the Problem
- II. Identification of Methane Sources
- III. Analytical Procedures/Methods
- IV. Data Interpretation Techniques

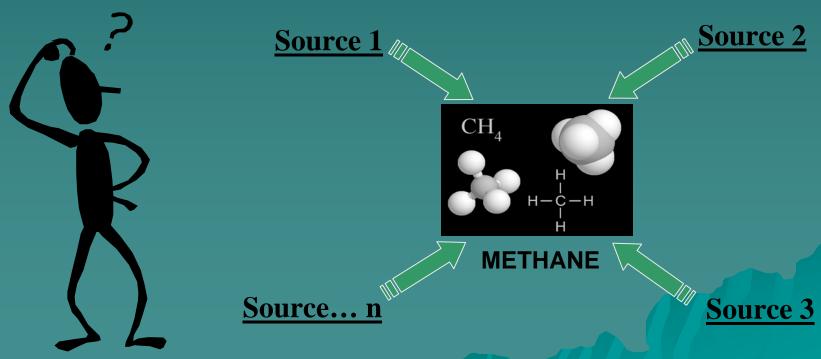
#### Case Studies in Orange County

Case I (Newport Terrace Landfill – Newport Beach, CA)
Case II (Cannery Street Landfill – Huntington Beach, CA)

- v. Background/Description of Case Studies
- vr. Workplan Implementation and Procedures
- VII. Results and Interpretation
- vIII. Conclusions

#### **Questions**

## I Introduction Stating the Problem



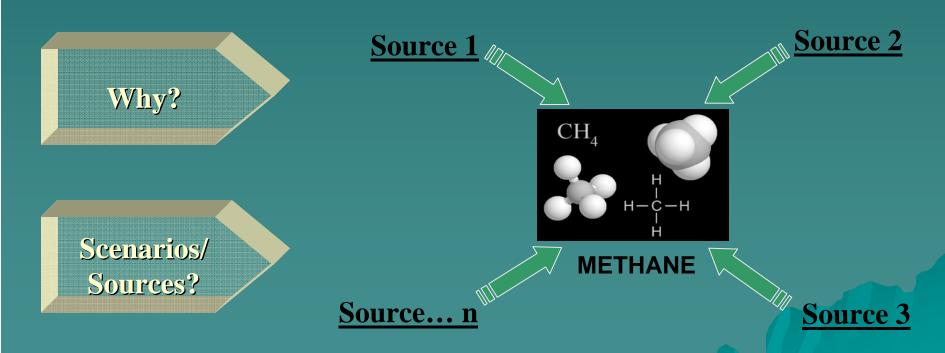
### **Problem Setting**

- Methane as greenhouse & explosive gas
- ◆ CH<sub>4</sub> Migration/Accumulation = Hazards
- ◆ LEA/CIWMB task = Gas Monitoring and Control (27 CCR, Section 20919)

Fugitive methane from **OTHER SOURCES** 

## Problem Setting (contd.)

Identification and correlation of methane releases to their source



### Why?

To determine Responsible Parties

◆ To determine LEA involvement

27 CCR 20913

 Adequate remediation design and implementation

"define the problem before the solution"

# III Identification of Methane Sources

What types of scenarios would the LEA encounter?





Common (Most Likely)

Rare (Not Likely)

#### Most Common Sources

Sources that generate CH<sub>4</sub> in high enough volumes and pressures to generate a migrating plume through soils

{FUGITIVE METHANE}

(Gas that you would see in your probes)

## Natural Gas Leaks (Pipeline Gas)

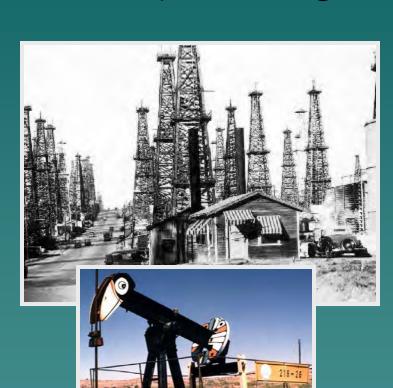


**Transportation Lines** 



Residential/Commercial Supply Lines

## Naturally Occurring Methane (Thermogenic/Petrogenic Gas)



Active/Abandoned Well Leaks



Natural Seepage (Underground Sources/Reservoirs)

## Other Biogas (Swamp/Marsh Gas)



Decay of Organic Matter (CH<sub>4</sub> Formation) &
Seepage
Through Soil



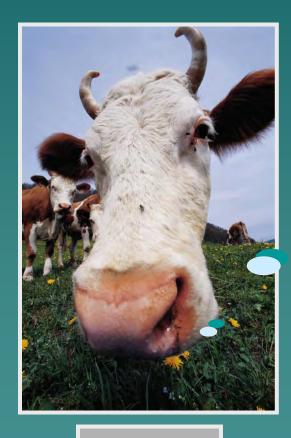
## Other Sources (Less Common/Not Likely)

- Low Potential Underground Migrating Plumes
- Not Enough Documented Data (CH<sub>4</sub> Migrating-Problems)

#### (Gas that you wouldn't likely see in your probes)

- Sewer Lines/Systems (accumulation @ enclosed spaces)
- Coal Mines
- Rice Fields
- Termites
- Oceans
- Livestock

### Factoid Livestock



A cow can belch up to ½ lb of CH4/Day



Sheep

Goats

Buffalo

Camels

Can Do It Too

Got Gas?

#### Factoid Livestock

97.1 Million Cattle in the US

<u>DO THE MATH!</u>

48 5 Million Ib CH. /Day



**Got Gas?** 

An untapped source of energy

### Not For Every Landfill in CA

Be Informed

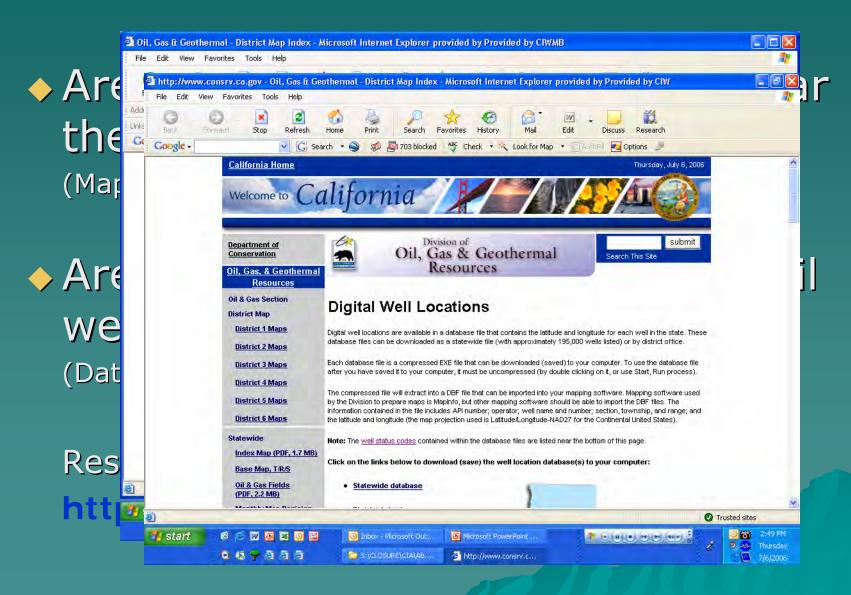
Do your Homework

There are tools

available!

San Diego

#### Oil & Gas Fields in CA



## Oil Production in CA 3 Major Regions

Kern County Region (69%)

 Los Angeles Basin Region (Central LA – Long Beach) (~10%)

Outer Continental Shelf (Offshore 10.2%)



# Forensics Correlating CH<sub>4</sub> Releases to their Source

## III Analytical Procedures/Methods

### Analytical Procedures/Methods

Fingerprinting Gas Releases (Gas Geochemistry)

#### Groups:

- 1. Looking at the Various Constituents
  - BTEX
  - $-H_2S$
  - $C_2 +$
  - Fixed Gases
  - Chlorinated Hydrocarbons/VOCs
  - Mercaptans

#### 2. Stable Isotope Composition

- Hydrogen Isotopic Ratio (<sup>2</sup>H/<sup>1</sup>H)
- Carbon Isotopic Ratio  $(^{13}C/^{12}C)$

#### 3. Radio Isotope Composition

- <sup>14</sup>C Concentration (Carbon Dating)
- 3H Concentration

#### 4. Measuring the Caloric Values

CH<sub>4</sub> BTU value

Not an Easy Task!

## Screening Process

Source x



**Gas Probe** 





Landfill



## Screening Process (contd.)

#### Formation Process

LFG

Swamp/Marsh Gas

Biogenic
Shallow depths/low temps
Anaerobic bacterial decomp
New Gas (0-100 years)

Naturally Occurring

Pipeline Gas

Thermogenic
High temps/depth/pressure
Thermal crack org. matter
Old Gas (Millions of years)

#### Elements for the ID of Sources

#### <u>LFG</u>

- ◆ CO<sub>2</sub> + CH<sub>4</sub> almost equal proportions
- VOCs/BTEX (trace)
- Low Oxygen
- $\bullet$  H<sub>2</sub>S (0-100 ppm) Note: 3-5% (US EPA)
- $\bullet$  C<sub>2</sub>-C<sub>5</sub> (trace)
- ◆ Unique Isotopic Fingerprinting (¹³C/¹²C and ²H/¹H)
- 14C Detected (Modern/New Gas)

## Elements for the ID of Sources (contd.)

#### **SWAMP/MARSH GAS**

- Easily mistaken with LFG
- Same formation process (biogenic/anaerobic)
- No VOCs/BTEX
- $\bullet$  No H<sub>2</sub>S
- Unique Isotopic Fingerprinting (<sup>13</sup>C/<sup>12</sup>C and <sup>2</sup>H/<sup>1</sup>H)

## Elements for the ID of Sources (contd.)

#### NATURALLY OCCURRING GAS

- BTEX (trace)
- No VOCs
- $\bullet$  H<sub>2</sub>S Low-High (0.1 2.5%)/(0-98%)
- $\bullet$  C<sub>2</sub>-C<sub>5</sub> (up to 20%)
- Unique Isotopic Fingerprinting (<sup>13</sup>C/<sup>12</sup>C and <sup>2</sup>H/<sup>1</sup>H)
- <sup>14</sup>C (Not Detected/Old Gas)

## Elements for the ID of Sources (contd.)

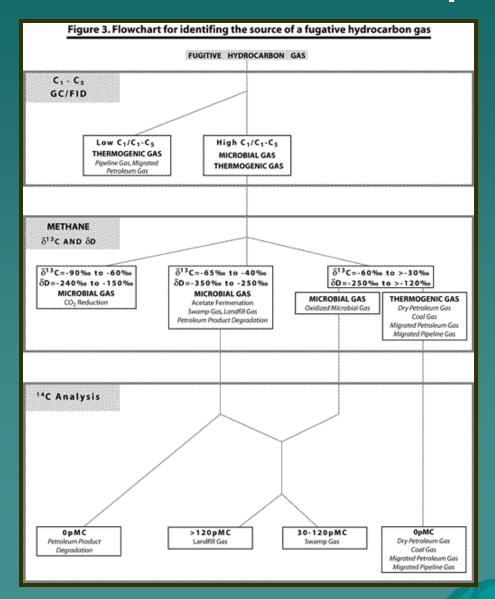
#### PIPELINE GAS

- Almost Pure CH<sub>4</sub> (80-90%)
- $\rightarrow$  Some C<sub>2</sub>-C<sub>5</sub> (%)
- Unique ID Element: Tracers (25-100 ppm)
   Mercaptans/Thiophene other Odorants

# Forensics Correlating CH<sub>4</sub> Releases to their Source

IV
Data Interpretation Techniques

### Data Interpretation



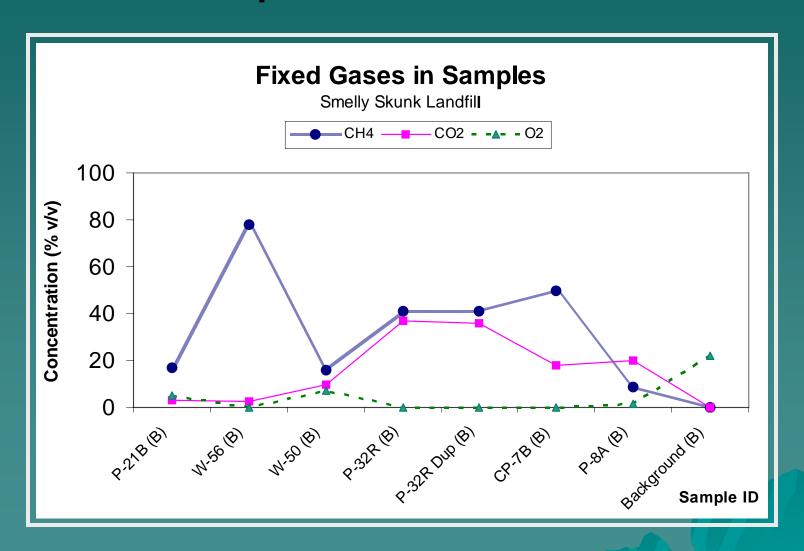
## Tier Evaluation Process

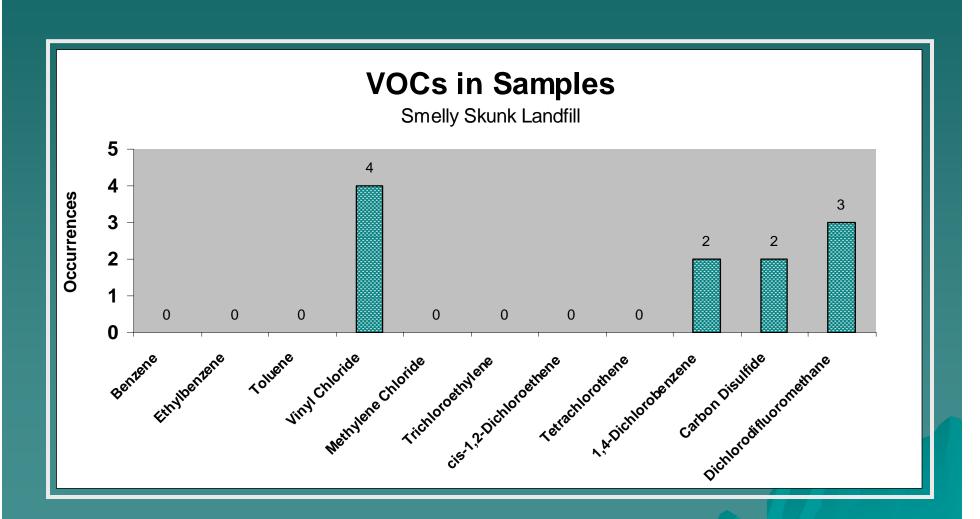
Source's
Unique
ID
ELEMENTS

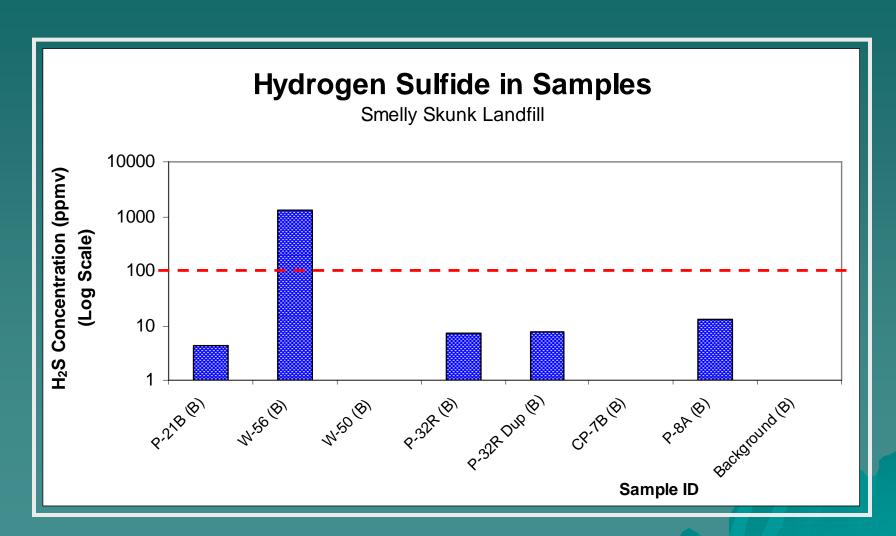


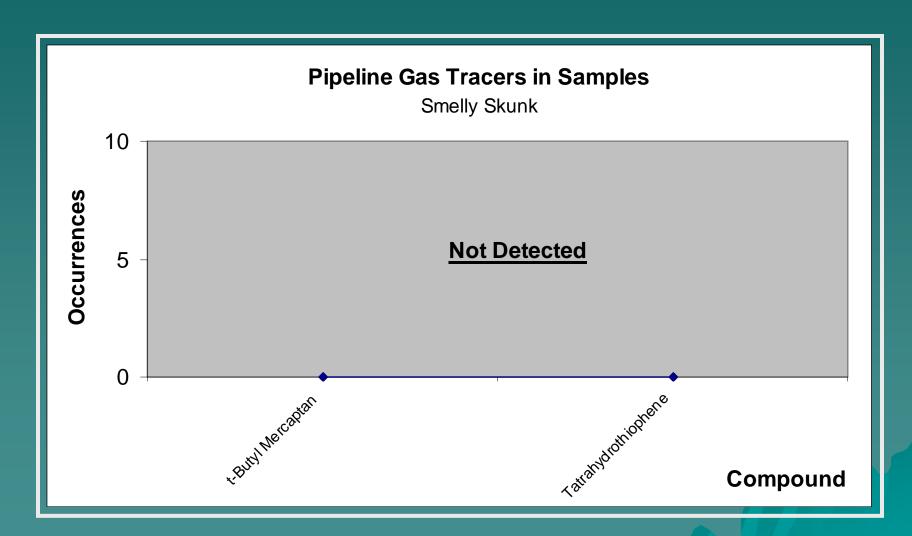


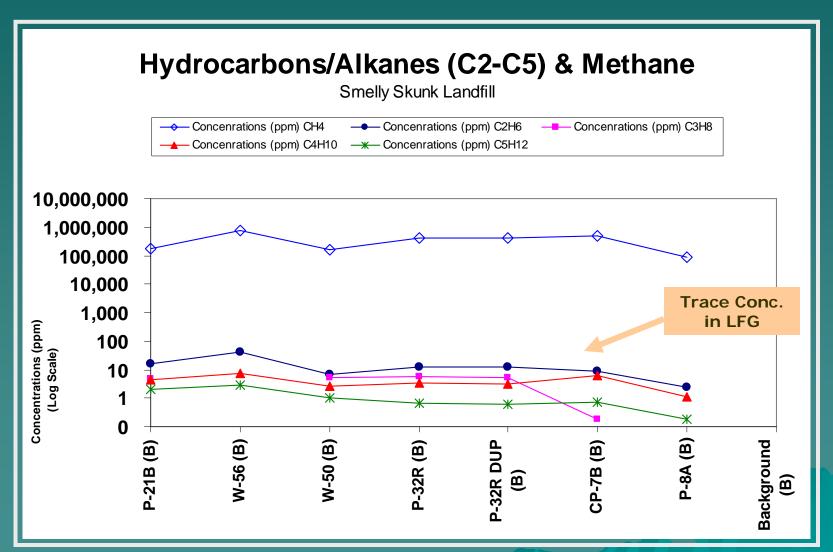
## Unique ID Elements

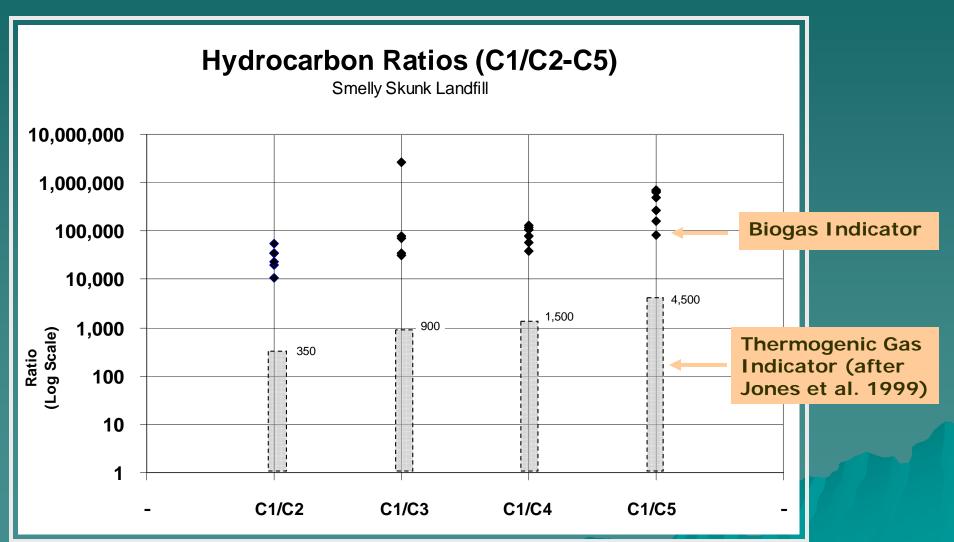












## Sophisticated Geochemistry to ID Sources

Stable Isotope Composition (Isotope Chemistry 101)

**Naturally Occurring Isotopes:** 

<sup>12</sup>C 98.89% <sup>13</sup>C 1.11%

Same Element
Different Atomic Weight

<sup>1</sup>H ~99.98% <sup>2</sup>H 0.0184% Same Element
Different Atomic Weight

### The Principle of Employing Isotopes

Distribution of Isotopes

Biogenic Methane Formation

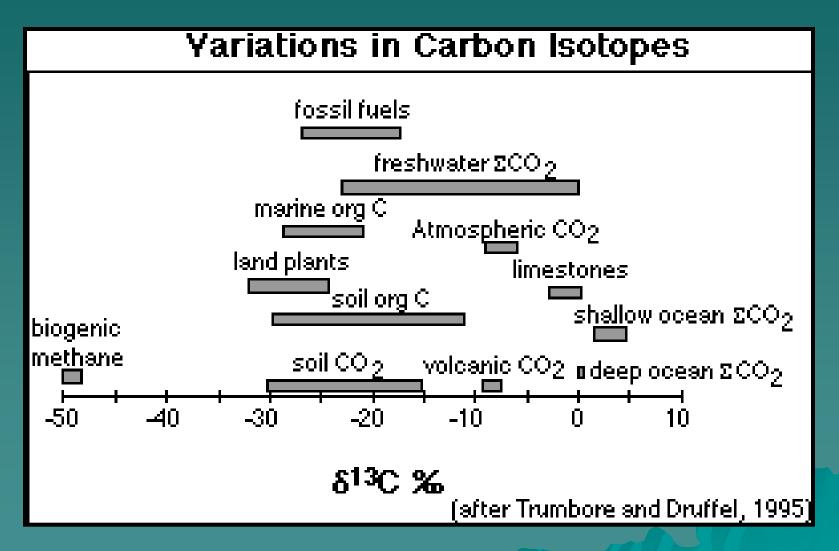
$$CH_3 -- |-- CO_2 - + H a$$

Bacterial Decomposition Light Isotopes <sup>12</sup>C & <sup>1</sup>H Preferentially Selected

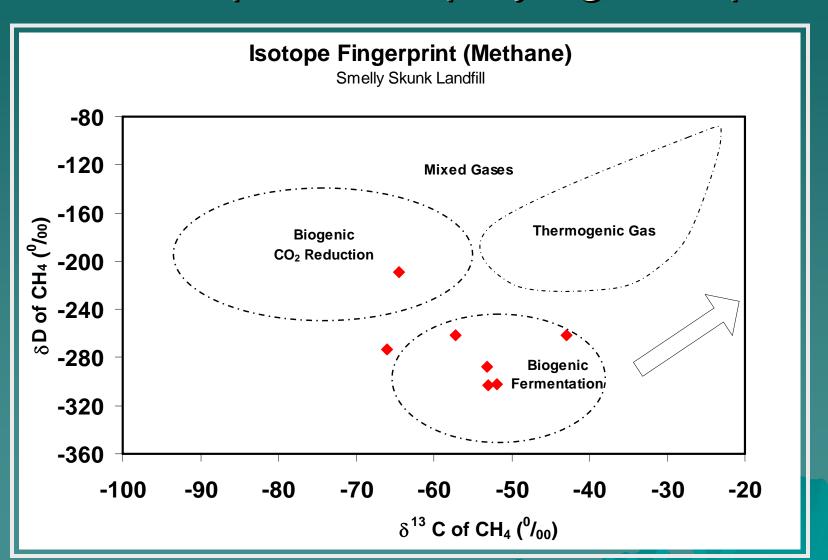
$$CO_2 + 8H \longrightarrow CH_4 + 2H_2O$$

<sup>13</sup>C/<sup>12</sup>C Ratio: Unique to the source <sup>2</sup>H/<sup>1</sup>H Ratio: Unique to the source

### The Principle of Employing Isotopes



#### The Principle of Employing Isotopes



## Radioisotope Composition

<sup>14</sup>C Concentration – Carbon Dating

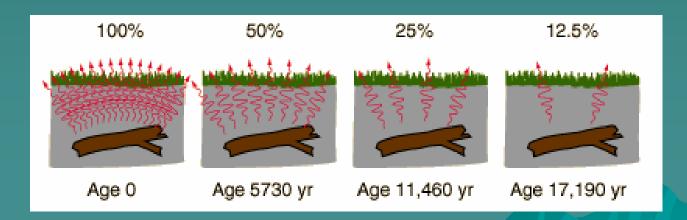
**Naturally Occurring Isotopes:** 

<sup>12</sup>C 98.89%

<sup>13</sup>C 1.11%

<sup>14</sup>C 0.000000001% (Radioactive)

<sup>14</sup>C Decays = Half-life 5730 yrs



How much

14C is
remaining
in the CH<sub>4</sub>?

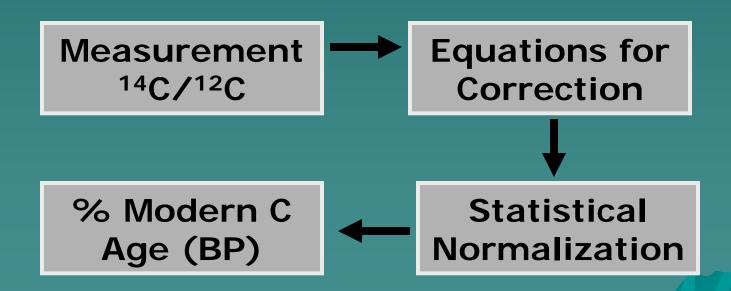
### <sup>14</sup>C Concentration – Carbon Dating

- Anything older than 60,000 years should have no detectable <sup>14</sup>C (thermogenic gas)
- ◆ If we detect ¹⁴C, it is good evidence that gas was generated less than 60,000 years (landfill gas)

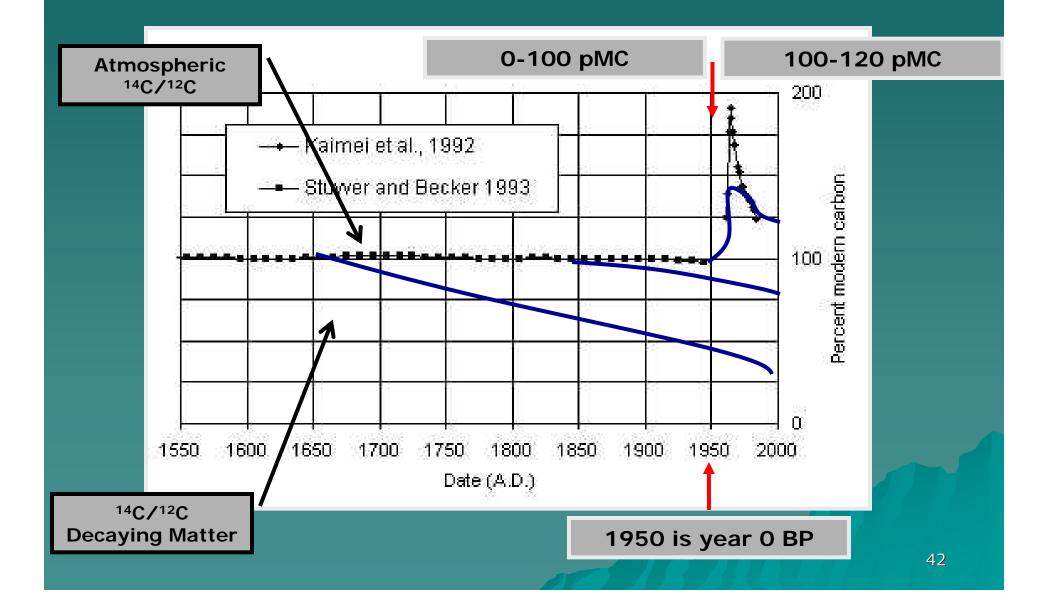
- 1.LFG should contain 14C
- 2. Thermogenic gas should not contain 14C

#### How is <sup>14</sup>C Measured

- Counting atoms/Accelerator MS
- AMS sophisticated UCI & Livermore)



### How is <sup>14</sup>C Measured



## Reporting <sup>14</sup>C Results

